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## SYNTHESIS AND EVALUATION OF NEW OXCARBAZEPINE ANALOGUES FOR ANTICONSULSANT ACTIVITY

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### Abstract

Epilepsy is one other most common ailment of man with a prevalence of approximately 1%. It is estimated that 50 millions person's worldwide may have this disorder. Although many are well controlled with available therapies, perhaps one quarter of the total continue to have seizures. Anticonvulsant drugs are the mainstay of epilepsy management and may have to be taken for life. In more than 20% of those affected, chronic intractable (refractory) epilepsy develops. This necessitates the use of combination therapy. But the use of these drugs in combination is plagued by cognitive impairment and drug interactions with the results that only about 10% of the patients with refractory epilepsy seem to benefit substantially from polypharmacy. Thus the new viable antiepileptic molecules are urgently needed. Oxcarbazepine has developed its efficacy and its improved safety profile and is now considered as epileptic drug of first choice. To potentiate the activity of oxcarbazepine for anticonvulsant activity the oxygen group of 10<sup>th</sup> position and NH<sub>2</sub> group of 5<sup>th</sup> position were changed with different groups such as methoxy, benzyl, piperazine, dimethylamine, methylamine which possess remarkable chemical and geometrical similarities for anti-convulsant activity. About six novel derivatives of oxcarbazepine were prepared. Their identification and characterization was done by TLC, IR, NMR spectroscopy.

**Keywords:** Epilepsy, Oxcarbazepine, Polypharmacy.

**Introduction:** The term epilepsy refers to a disorder of brain function characterized by the periodic and unpredictable occurrence of seizures<sup>1,2</sup>. The term "seizures" refer to a transient alteration of behaviour due to abnormal excessive, hyper synchronous discharges from aggregate of CNS neurons.

Seizure can be

- a) Non epileptic –when evoked in a brain by treatment such as electric shock or chemical convulsants.
- b) Epileptic –when occur without evident provocation.

The epilepsy is common and frequently devastating disorder, affecting approximately 0.5% of the population. More than 40 distinct forms of epilepsy have been identified. The incidence increases again epilepsy begins before the age of 18 in over 75% population.

Seizures the characteristic event in epilepsy, is associated with the episodic high frequency discharges of impulses by a group of neurons in the brain. What starts as a local discharge may then spread to other areas of the brain. The site of primary discharge and extent of its spread determines the symptoms that are produced, which ranged from a brief lapse of attention to a full conclusive fit lasting for several minutes.

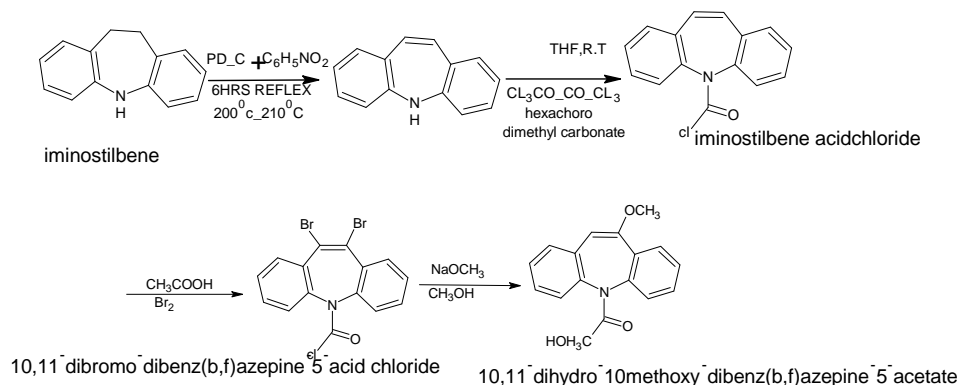
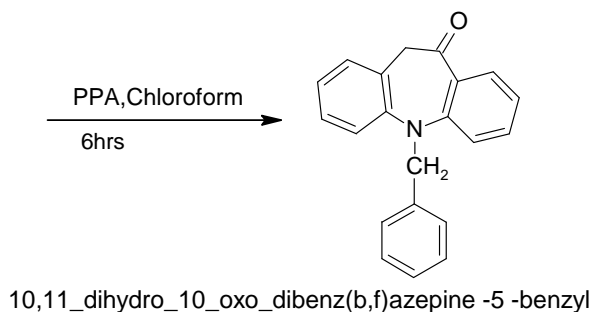
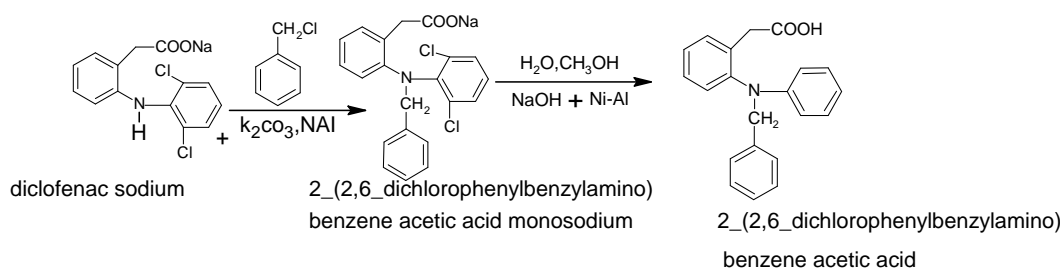
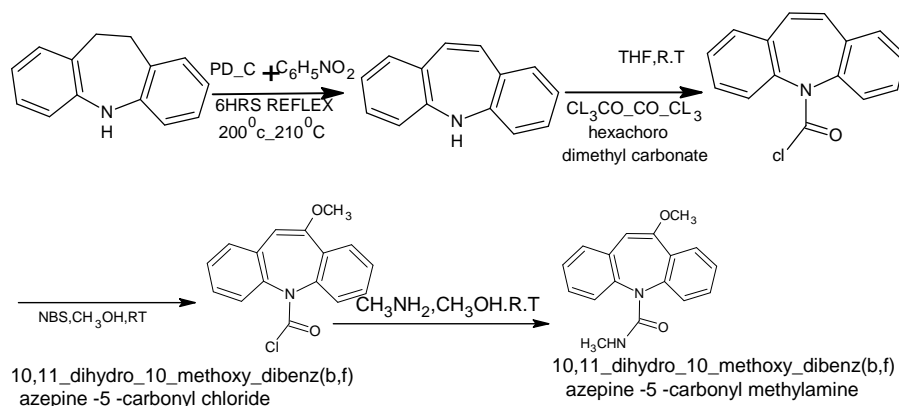
The particular symptoms produced depend on the function of the region of the brain that is affected thus involvement of the hypothalamus causes peripheral autonomic discharge and involvement of the reticular formation in the upper brain stem leads to loss of consciousness.

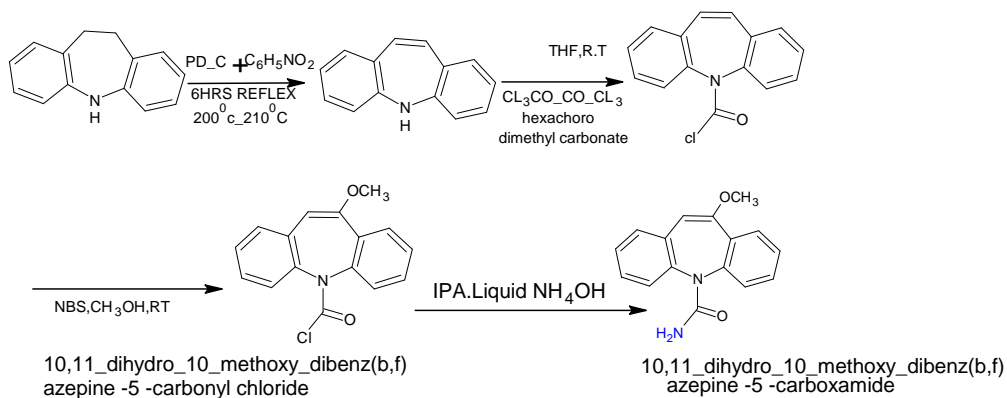
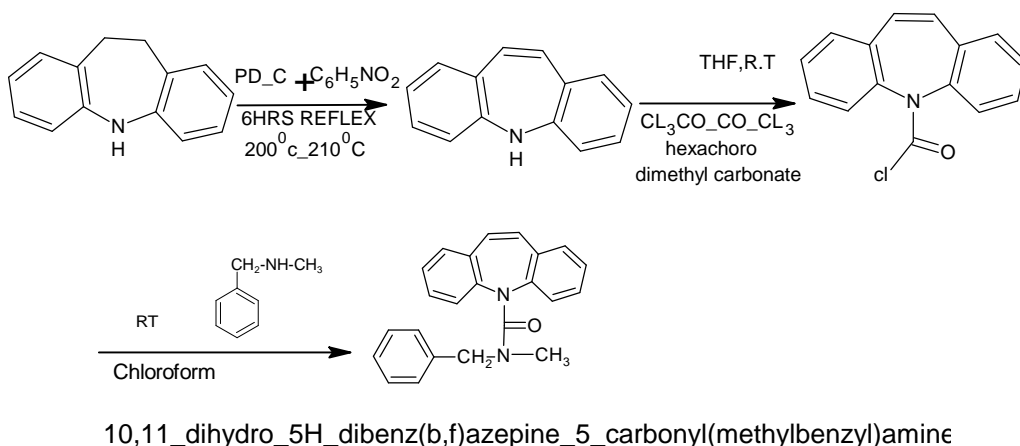
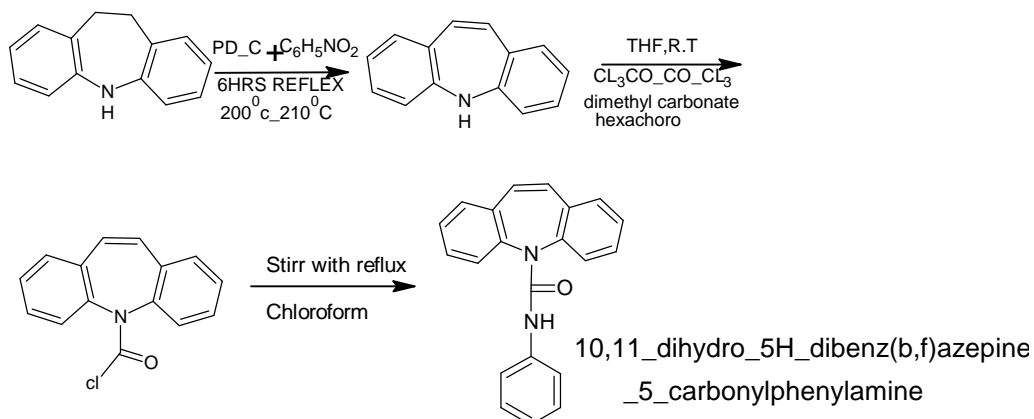
### **Objectives of the Study**

Oxcarbazepine and its analogues are highly lipophilic thus easily pass the BBB and reach the site of action<sup>3</sup>. Hence, an attempt was made to prepare oxcarbazepine analogues by modified route and screen for anticonvulsant activity.

### **The Main Objectives of the Study**

- ✓ Attempt was made to synthesize oxcarbazepine analogues through different routes that may give good yield with acceptable purity
- ✓ Costly raw materials were avoided to make the process economical.
- ✓ Complicated procedures were avoided to achieve simpler techniques.
- ✓ Characterization of synthesized oxcarbazepine analogues was made by TLC, IR, and NMR techniques.
- ✓ Pharmacological evaluation of oxcarbazepine by maximal electrical shock method which produce a selective anticonvulsant activity was studied.

**SYNTHESIS OF ANALOGUE -1 (Derivative -1):****SYNTHESIS OF DERIVATIVE-2:****SYNTHESIS OF DERIVATIVE -3**

**SYNTHESIS OF DERIVATIVE -4****SYNTHESIS OF DERIVATIVE -5****SYNTHESIS OF DERIVATIVE -6****PHYSICAL PROPERTIES (D1-D6):**

The physical properties of all the synthesized derivatives are given in Table – 1.

**Table No-1: The physical properties of all the synthesized derivatives.**

PROPERTIES	OBSERVATIONS					
	Derivative-D1	Derivative-D2	Derivative-D3	Derivative-D4	Derivative-D5	Derivative-D6
Description	It is yellow and odourless	Yellow crystals and odourless	Brown, solid and odourless	Dark brown and odourless	Brown crystals and odourless	White crystals and odourless
Solubility	Acetone, Methanol, Ethanol, Chloroform, sparingly soluble in toluene	Ehyl acetate, Choloroform, Ethanol, Methanol Sparingly soluble in PE	EA, Chloroform, Methanol, Ethanol, Acetone, Sparingly soluble in Toluene	EA, Acetone, Methanol, Toluene, Sparingly soluble or insoluble in water	EA, Acetone, chloroform, Methanol IPA and insoluble in water	EA, Acetone, chloroform, Methanol IPA, Toluene and insoluble in water
Recrystallized by	Column Chromatography	Column Chromatography	Ethanol	Column Chromatography	Ethanol	Column Chromatography
Melting Point	130 <sup>0</sup> C-132 <sup>0</sup> C	148 <sup>0</sup> C-150 <sup>0</sup> C	128 <sup>0</sup> C-130 <sup>0</sup> C	122 <sup>0</sup> C-125 <sup>0</sup> C	105 <sup>0</sup> C-110 <sup>0</sup> C	138 <sup>0</sup> C-142 <sup>0</sup> C
% of Yield (In final stage)	90	84	80	79	90	86

**I.R SPECTRA (D1-D6):**

The IR spectra of the synthesized derivatives are given in Table – 2, 3 & 4.

**Table-2: The IR spectra of the synthesized derivatives 1 and 2.**

<b>IR Spectrum D1</b>	Wave Numbers	<b>IR Spectrum D1</b>	Wave Numbers
Groups	(CM <sup>-1</sup> )	Groups	(CM <sup>-1</sup> )
C=O Str of ester	1705	C=O Str of ester	1727
C=O Str of Methoxy group	1439	Aromatic C=C Str	1611-1490
C=C Str of Aromatic ring	1589	Aromatic CH Str	3030
Aromatic out of plane CH deformation vibrations	700-800	Aromatic out of plane CH deformation vibrations	700-800
CH Str of Alkane	2390		

**Table –3: The IR spectra of the synthesized derivatives 3 and 4.**

<b>IR Spectrum D3</b>	Wave Numbers	<b>IR Spectrum D4</b>	Wave Numbers
Groups	(CM <sup>-1</sup> )	Groups	(CM <sup>-1</sup> )
C=O Str of amide group	1661	C=O Str of amide group	1672
NH Str of amide group	3366	Asymmetric & Symmetric Str of NH <sub>2</sub> group of amide	3476.5 3348
Aliphatic CH Str of azepine	2930	C=C Str of Aromatic ring	1583 & 1479.3
Aromatic C=C Str	1473.5	C-H deformation vibration	756
C-O Str of methoxy group	1111.9		

**Table –4: The IR spectra of the synthesized derivatives 5 and 6.**

<b>IR Spectrum D5</b>	Wave Numbers	<b>IR Spectrum D6</b>	Wave Numbers
Groups	(CM <sup>-1</sup> )	Groups	(CM <sup>-1</sup> )
C=O Str of amide group	1647	C=O Str of amide group	1663
CH <sub>2</sub> Str of benzyl group	2930	NH Asymmetric & Symmetric Str of Amide group respectively	3421 3333
Aromatic C=C Str	1400	Aromatic C=C Str	1525.8
C-H deformation vibration	741.6		

**NMR SPECTRA (D1-D6):**

The NMR spectra of the synthesized derivatives are given in Table – 5 & 6.

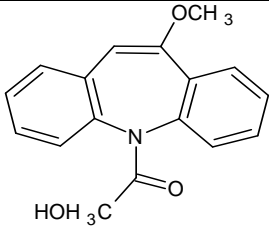
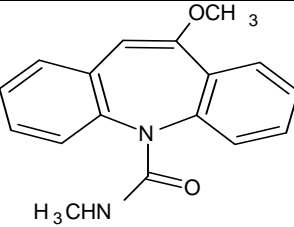
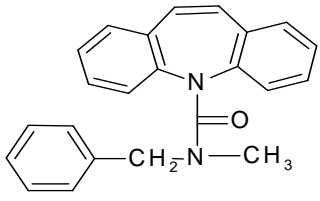
**Table-5: The NMR spectra of the synthesized derivatives 1, 2 and 3.**

NMR Spectrum D1	NMR Spectrum of D2	NMR Spectrum D3
Two sharp methoxy protons at 3.7 $\delta$ ppm	The azepine proton of 10 <sup>th</sup> position and benzyl proton of CH <sub>2</sub> group merged and gave quintet at 3.4 $\delta$ ppm	Methoxy protons and Methyl protons at 3.4 and 2.9 $\delta$ ppm respectively
At 6.7 $\delta$ ppm singlet proton of azepine ring system		NH proton of amide group at 5.2 $\delta$ ppm
Between 7 to 8 $\delta$ ppm aromatic protons	Aromatic protons between 7 to 8 $\delta$ ppm	Aromatic proton between 7 to 8 $\delta$ ppm
		Proton of azepine ring at 4.2 $\delta$ ppm

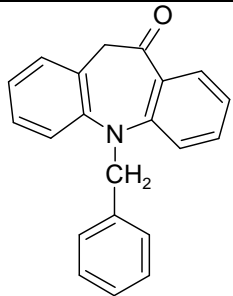
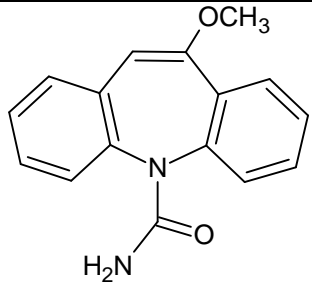
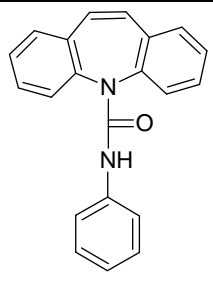
**Table-6: The NMR spectra of the synthesized derivatives 4, 5 and 6.**

NMR Spectrum of D4	NMR Spectrum of D5	NMR Spectrum of D6
Methoxy protons peak at 3.4 $\delta$ ppm	Methyl protons at 2.5 $\delta$ ppm	NH protons at 6.3 $\delta$ ppm
Amide protons at 4.7 and 5 $\delta$ ppm respectively	Methylene protons at 4.3 $\delta$ ppm	Aromatic protons between 7 to 8 $\delta$ ppm
Aromatic protons between 7 to 8 $\delta$ ppm	Aromatic protons between 7 to 8 $\delta$ ppm	Azepine ring protons at 6.9 $\delta$ ppm
Azepine ring protons at 6.9 $\delta$ ppm	Azepine ring protons at 6.9 $\delta$ ppm	

According to the IR and NMR spectral data obtained the structures of the compounds may be as follows:

DERIVATIVE-D1	DERIVATIVE-D2	DERIVATIVE-D3
		

10,11-dihydro-10 methoxy - 5H -dibenz(b,f)azepine-5- acetate.	10,11-dihydro-10-methoxy- 5H-dibenz (b,f) azepine-5 - carbonyl methyl amine.	10,11-dihydro-5H-dibenz(b,f) azepine-5 –carbonyl (methyl benzyl) amine.
Attributed Molecular formula: C <sub>17</sub> H <sub>15</sub> NO <sub>3</sub>	Attributed Molecular formula: C <sub>17</sub> H <sub>16</sub> NO <sub>2</sub>	Attributed molecular formula: C <sub>23</sub> H <sub>21</sub> N <sub>2</sub> O

DERIVATIVE-D4	DERIVATIVE-D5	DERIVATIVE-D6
		
10,11-dihydro-10-oxo- dibenz(b,f)azepine-5- Benzyl	10,11-dihydro- 0-methoxy -5H-dibenz(b,f) azepine-5-carboxamide	10,11-dihydro-5H-dibenz (b,f) azepine-5-carbonyl (phenyl) amine
Attributed Molecular formula: C <sub>21</sub> H <sub>17</sub> NO	Attributed Molecular formula: C <sub>15</sub> H <sub>14</sub> NO <sub>2</sub>	Attributed molecular formula: C <sub>21</sub> H <sub>17</sub> N <sub>2</sub> O

#### Anticonvulsant activity of synthesized oxcarbazepine analogues by maximal electroshock method<sup>4</sup>.

Institutional Animal Ethical Committee (IAEC) approval was taken before performing the animal studies.

In maximal electroshock (MES) electroshock is applied through the ear electrodes. The MES convulsions are divided into five phases such as

1. Tonic flexion
2. Tonic extensor
3. Clonic convulsion



4. Stupor and

5. Recovery or death

A substance is to possess anticonvulsant property if it reduces or abolished the extensor phase of MES convulsion.

**Procedure:** Swiss albino rats of either sex (170-250gm) were weighed and divided into 8 groups of 5 animals each.

Of the 8 groups one group was used as control, to test the effect of MES alone and one group was used as STD

administered with oxcarbazepine 14mg/kg<sup>1</sup> and remaining groups were used for synthesized oxcarbazepine analogues

The suspension of the drugs were prepared in 0.5% guar gum solution in distilled water. To the control animals MES

was given and time for onset of clonic convulsions was recorded. To the other groups 14mg/kg<sup>9</sup> of oxcarbazepine was

given through oral route and after 1 hour MES was given and delay in the onset of action were recorded.

To the remaining six groups, synthesized oxcarbazepine analogues were given through oral route and after one hour,

the onset of convulsions were recorded. Results were statistically compared against the control.

Group no	Body Wt	Treatment (dose)	Time(sec) in various phase of convulsions				
			flexion	Extensor	clonus	Stupor	Ror D
I	189	Control	6	9	3	90	R
	190		5	8	4	110	D
	195		4	11	2	96	R
	192		5	10	4	102	R
	193		7	9	2	98	D
II	185	Oxcarbazepine 14mg/kg	2	6	2	60	R
	180		2	3	1	58	R
	175		0	3	1	71	R
	183		2	1	0	42	R
	182		0	2	2	48	R
III	205	D1	2	8	2	80	R
	208		3	7	3	83	R

	207	<b>(14mg/kg)</b>	0	8	1	78	R
	203		4	8	2	75	R
	210		1	9	1	79	R
IV	210	<b>D2 (14mg/kg)</b>	3	6	4	76	R
	215		4	7	3	78	R
	213		3	8	2	84	R
	214		2	6	1	90	R
	215		0	8	2	70	R
V	210	<b>D3 (14mg/kg)</b>	3	9	3	90	R
	212		0	8	4	79	R
	214		1	6	3	80	R
	213		3	5	2	75	R
	216		2	7	1	74	R
VI	233	<b>D4 (14mg/kg)</b>	0	5	3	80	R
	236		2	7	4	79	R
	240		1	6	3	70	R
	235		1	9	2	78	R
	241		1	7	2	70	R
VII	170	<b>D5 (14mg/kg)</b>	4	10	5	100	R
	173		5	11	6	110	D
	175		7	09	7	105	R
	178		6	10	8	103	R
	180		3	12	13	115	D

VIII	210	<b>D6 (14mg/kg)</b>	1	6	3	80	R
	203		4	7	4	79	R
	207		3	8	3	70	R
	208		0	6	2	78	R
	205		4	8	1	70	R

R = Recovery, D = Death

The following table indicates the effects of synthesized oxcarbazepine and its analogues on the extensor phase of convulsions in albino rats.

S. No	Treatment dose (mg/kg)	Mean $\pm$ SEM	P value
1	Control	4.60 $\pm$ 0.748***	-
2	Oxcarbazepine (14mg/kg)	3.00 $\pm$ 0.836***	0.0002
3	D1 (14mg/kg)	8.00 $\pm$ 0.316*	0.0479
4	D2 (14mg/kg)	7.00 $\pm$ 0.447*	0.0076
5	D3 (14mg/kg)	7.00 $\pm$ 0.707*	0.0249
6	D4 (14mg/kg)	6.80 $\pm$ 0.663*	0.0145
7	D5 (14mg/kg)	10.40 $\pm$ 0.509*	0.2029
8	D6 (14mg/kg)	7.00 $\pm$ 0.447**	0.0076

Statistics: Unpaired 't' test n=5, \*P<0.05, \*\*P<0.01, \*\*\*P<0.001

## Results

The synthesized oxcarbazepine analogues were compared with that of control for anticonvulsant activity. Among the six synthesized oxcarbazepine analogues five of them were found to be active as they reduced time of extensor phase

as compared to control and the values are statistically significant as compared to control. Remaining one analogue D5 did not show any anticonvulsant activity in the parameter tested.

## Discussion and Conclusion

Oxcarbazepine has developed its efficacy and its improved safety profile and is now considered as epileptic drug of first choice. To potentiate the activity of oxcarbazepine for anticonvulsant activity the oxygen group of 10<sup>th</sup> position and NH<sub>2</sub> group of 5<sup>th</sup> position were changed<sup>8</sup> with different groups such as methoxy, benzyl, piperazine, dimethylamine, methylamine which possess remarkable chemical and geometrical similarities for anti-convulsant activity. About six novel derivatives of oxcarbazepine were prepared. Their identification and characterization was done by TLC, IR, NMR spectroscopy. The synthesized oxcarbazepine analogues were screened for anticonvulsant activity by maximal electrical shock method (MES). The animal studies proved that D5 is not significant in its activity. D2 and D6 are said to be more significant. The animal studies have been performed with respect to oxcarbazepine as standard, using 14mg/kg<sup>14,15</sup>. The animal studies using 14mg/kg have proved that D1, D2, D3, D4, & D6 are significant in their activity. However LONG TERM TOXICITY STUDIES have to be carried out before any conclusion is drawn.

## References

1. Goodman and Gilman's *The Pharmacology basis of therapeutics* 9<sup>th</sup> edn; 1996 pp 521-547, Mc Graw Hill Inc, New York.
2. Rang, H.P, Dale: *Pharmacology*, Drugs used in treating motor disorders: Epilepsy history, Parkinsonism and spasticity. 2<sup>nd</sup> edn. 1991. pp 684-696.
3. *Burger's Medicinal Chemistry and Drug Discovery*, 5<sup>th</sup> edn, vol3: Therapeutics agent's pp 175-260.
4. Kulkarni, S.K. Experiment 4.11. In: *Hand book of Experimental pharmacology*, 3<sup>rd</sup> edn, 1999 pp 131-133.vallabh prakashan, Delhi.
5. Oliver lohse and Ulrich et al: New synthesis of oxcarbazepine via remote methylation of protected N-o-tolyl-anthanimide derivatives: *Tetrahydron letters* ,vol 42,2001 pp 385-389
6. Roland Heckendron: Synthesis of trans -10-11-dihydro-10-11-dihydroxy-5H-dibenz (b, f) azepine-5-carboxamide, a major metabolite of carbamazepine, In: *Helvetica chemica acta.*, vol.70, 1987, pp-1955- 1962.

7. Tomio Ohata, Naoki miyata et al; studies on the oxidation of N-substituted-dibenz (b,f) azepine. II Synthesis and reaction of 5H- dibenz (b,f) azepine -10-11-oxide, In: *Chem. Pharm Bull*; vol 32 (10) 1984 pp 3857-3865.
8. Milanese; Alberto, United States Patent No; 5808058 1998.
9. Wamil et al.; *European Journal of Pharmacology* vol. (2), 1994 pp 301-308.
10. L.J. Kricka & A. Ledwith: Dibenz (b,f) azepines and Ring systems In: *Chemical Reviews*, vol 74. 1974 pp 102-123.
11. Wilson and Gisvold's *The Text Book of Organic Medicinal and Pharmaceutical Chemistry*, 9<sup>th</sup> edn 1991, 11, pp 334-393.
12. Vogel's *Text Book of Practical Organic Chemistry* 4<sup>th</sup> edn. London ELBS: 1994, pp 932-933.
13. Beckett, A H and Stenlake, J. B. Thin layer chromatography In: *Practical Pharmaceutical Chemistry*, 1960 pp 473.
14. Silverstein, R. M, *Infrared Spectroscopy: In Identification of Organic compounds* 5<sup>th</sup> edn, New york: John wiley and Son's: 1961, pp 103-128.
15. Kemp, W. Proton NMR spectra. In: *Chemistry a Multinuclear Introduction*, 2<sup>nd</sup> edn, New York: Mc millan, 1986; pp 211-214.

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