



ISSN: 0975-766X  
CODEN: IJPTFI  
Research Article

Available Online through  
[www.ijptonline.com](http://www.ijptonline.com)

## SYNTHESIS OF 7- SUBSTITUTED-4-HYDROXYQUINOLIN-2(1H)-ONE FOR ANTI-BACTERIAL AND ANTI-FUNGAL ACTIVITY

\*Subba Rami Reddy SR, Subbi Reddy E, Suryanarayana Rao.V

Department of Chemistry, Sri Krishnadevaraya University, Anantapur 515003 India.

Email: [srsubbaramireddy@gmail.com](mailto:srsubbaramireddy@gmail.com)

Received on 06-04-2012

Accepted on 28-04-2012

### Abstract

The present work is carried out for the synthesis of 4-Hydroxy-1*H*-quinolin-2-ones derivatives with well established procedures. A new series of 4-Hydroxy-1*H*-quinolin-2-ones derivatives were synthesized and the structures of these compounds were confirmed by IR, HNMR and Mass Spectroscopy. The title compounds were screened for antibacterial activity and antifungal activity.

**Keywords:** 7- Substituted-4-hydroxyquinolin-2(1H)-one, IR, HNMR, Antibacterial and Antifungal activity.

### Introduction

Many 4-hydroxy quinolone compounds are useful intermediates for biologically active substances [1-3] and many industrial products such dye stuffs [4, 5] and herbicides [6, 7]. Recent reports [8, 9] describe novel class of 4-hydroxy quinolone derivatives, i.e., 4-hydroxy-3-arylquinoline-2(1H)-one and 4-hydroxy-3-nitroquinoline-2(1H)-one as potent and selective N-methyl-D-aspartate (NMDA) receptor glycine site antagonist after oral administration. Many of these compounds have been derived from the 4-hydroxyquinoline-2(1H)-one nucleus. A variety of routes for the synthesis of 4-hydroxy quinolone compounds have been reported in the literature [10-15].

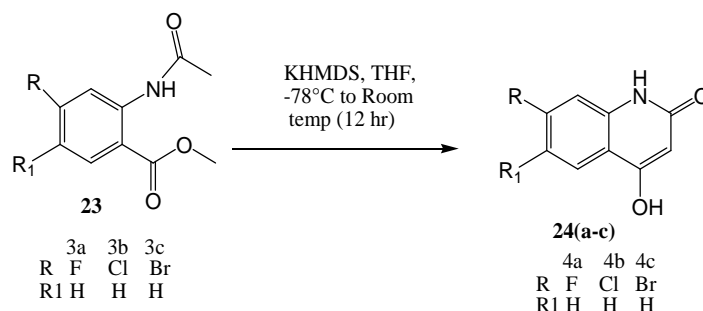
### Materials and Methods

#### Synthesis of 7-fluoro-4-hydroxyquinolin-2(1H)-one:

To a solution of methyl 2-acetamido-4-fluorobenzoate (3.0 gm, 14.218 mmoles) in THF (40 mL) was added slowly Potassium bis (trimethylsilyl) amide (commonly abbreviated as KHMDS) (42.65 mL, 21.327 mmoles) (0.5 M solution in toluene) at -78°C. The reaction mixture was then allowed to room temperature and stirred at room temperature for 12 hours under nitrogen atmosphere. The reaction was monitored by TLC, after completion of reaction the reaction mixture was quenched with water (50 mL).The aqueous layer was washed with

Ethyl acetate (50 mL), separate the layers, taken aqueous layer and cooled to 10°C and slowly pH adjusted 2.5 -3.5 with 5N HCl, to give a white solid, which was collected by filtration and dried in vacuum to give 7-fluoro-4-hydroxyquinolin-2(1H)-one as a white solid (Figure-1).

**Figure: 1 Synthesis of 7- Substituted-4-hydroxyquinolin-2(1H)-ones from Methyl 2-acetamido-4- substituted benzoates.**



**3a methyl 2-acetamido-4-fluorobenzoate, 3b methyl 2-acetamido-4-chlorobenzoate  
3c methyl 2-acetamido-4-bromobenzoate, 4a 7-fluoro-4-hydroxyquinolin-2(1H)-one  
4b 7-chloro-4-hydroxyquinolin-2(1H)-one, 4c 7-bromo-4-hydroxyquinolin-2(1H)-one**

#### Synthesis of 7-chloro-4-hydroxyquinolin-2(1H)-one and Synthesis of 7-bromo-4-Hydroxyquinolin-2(1H)-one:

According to the same procedure (Synthesis of 7-fluoro-4-hydroxyquinolin 2(1H)-one) 7-chloro-4-hydroxyquinolin-2(1H)-one Compound was obtained from methyl 2-acetamido-4-chlorobenzoate 75% yield, White solid, 7-bromo-4-hydroxyquinolin-2(1H)-one Compound was obtained 76% yield from methyl 2-acetamido-4-bromobenzoate, White solid.

#### Anti bacterial activity:

The antibacterial activity of synthesized compounds was studied by the disc diffusion method against the following pathogenic organisms. The gram-positive bacterial screened were Bacillus subtilis ATCC 6633 and gram negative bacterial screened were Escherichia coli ATCC. The synthesized compounds were used at the concentration of 50µg, 100µg and 150µg using DMSO as a solvent. The Streptomycin sulphates were used as a standard. (Raghavendra Institute of pharmaceutical Education and Research (RIPER) Anantapur, Andhra pradesh).

#### Anti fungal activity:

The antifungal activities of synthesized compounds were studied by disc diffusion method against the organisms of Aspergillus Niger MTCC F2723. Compounds were treated at the concentrations of 50 µg/ml, 100

$\mu\text{g/ml}$  and  $150 \mu\text{g/ml}$  using DMSO as a solvent. The standard used was Amphotericin  $50 \mu\text{g/ml}$  against both the organisms.

## Results and Discussion

### Characterization of 7-fluoro-4-hydroxyquinolin-2(1H)-one:

From the above scheme, synthesis of 7-fluoro-4-hydroxyquinolin-2(1H)-one **was** characterized by  $^1\text{H}$  NMR and Mass. Yield = 2.0 gm, (78.53 %), melting Point =  $290\text{-}291.5^\circ\text{C}$  and HPLC purity = 98.80 %. Thus, its IR (in KBr) showed a strong peak at  $3604 \text{ cm}^{-1}$  due to -OH grouping,  $1686 \text{ cm}^{-1}$  due to  $-\text{NH}-\text{C}=\text{O}$  grouping and  $1630 \text{ cm}^{-1}$  due to  $-\text{C}=\text{O}$  grouping. Its  $^1\text{H}$  NMR (DMSO- $d_6$ /TMS):  $\delta$  5.70-5.80 (s, 1H), 7.20-7.25 (d, 1H), 7.40-7.50 (s, 1H), 7.70-7.80 (d, 1H), 11.10-11.20 (s, 1H), 11.40-11.45 (s, 1H) and its ESI mass spectrum showed molecular ion peak at 180.22 ( $\text{M}^++1$ ) corresponding to the molecular mass of 179.34 ( $\text{M}^+$ ).

### Characterization of 7-chloro-4-hydroxyquinolin-2(1H)-one:

Synthesis of 7-chloro-4-hydroxyquinolin-2(1H)-one was characterized by  $^1\text{H}$  NMR and Mass. Melting Point =  $300\text{-}301^\circ\text{C}$  and HPLC purity = 98.10 %. Thus, its IR (in KBr) showed a strong peak at  $3604 \text{ cm}^{-1}$  due to -OH grouping,  $1686 \text{ cm}^{-1}$  due to  $-\text{NH}-\text{C}=\text{O}$  grouping and  $1630 \text{ cm}^{-1}$  due to  $-\text{C}=\text{O}$  grouping. Its  $^1\text{H}$  NMR (DMSO- $d_6$ /TMS):  $\delta$  5.72-5.82 (s, 1H), 7.25-7.35 (d, 1H), 7.45-7.55 (s, 1H), 7.75-7.85 (d, 1H), 11.0-11.10 (s, 1H), 11.20-11.30 (s, 1H) and its ESI mass spectrum showed molecular ion peak at 196.12 ( $\text{M}^++1$ ) corresponding to the molecular mass of 195.14 ( $\text{M}^+$ ).

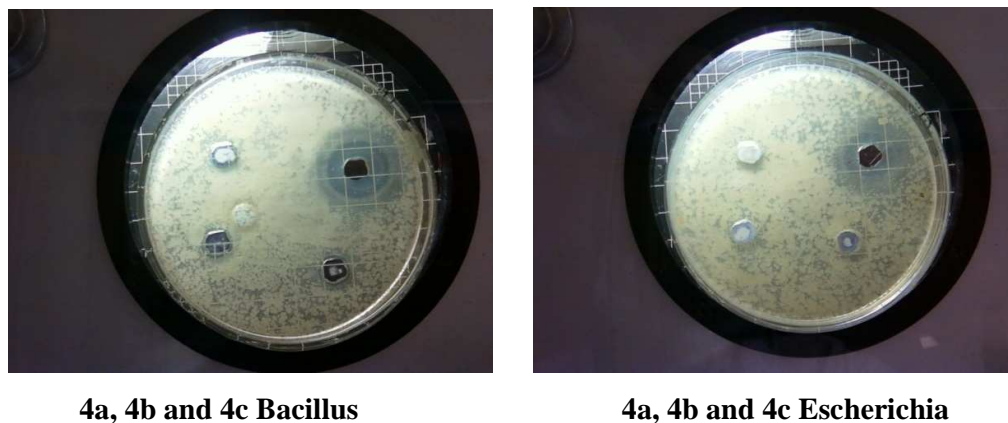
### Characterization of 7-bromo-4-hydroxyquinolin-2(1H)-one:

Synthesis of 7-bromo-4-hydroxyquinolin-2(1H)-one was characterized by  $^1\text{H}$  NMR and Mass. Melting Point =  $320\text{-}321^\circ\text{C}$  and HPLC purity = 98.10 %. Thus, its IR (in KBr) showed a strong peak at  $3604 \text{ cm}^{-1}$  due to -OH grouping,  $1686 \text{ cm}^{-1}$  due to  $-\text{NH}-\text{C}=\text{O}$  grouping and  $1630 \text{ cm}^{-1}$  due to  $-\text{C}=\text{O}$  grouping. Its  $^1\text{H}$  NMR (DMSO- $d_6$ /TMS):  $\delta$  5.75-5.82 (s, 1H), 7.35-7.37 (d, 1H), 7.46-7.57 (s, 1H), 7.78-7.90 (d, 1H), 11.1-11.20 (s, 1H), 11.25-11.35 (s, 1H) and its ESI mass spectrum showed molecular ion peak at 239.9 ( $\text{M}^++1$ ) corresponding to the molecular mass of 239.0 ( $\text{M}^+$ ).

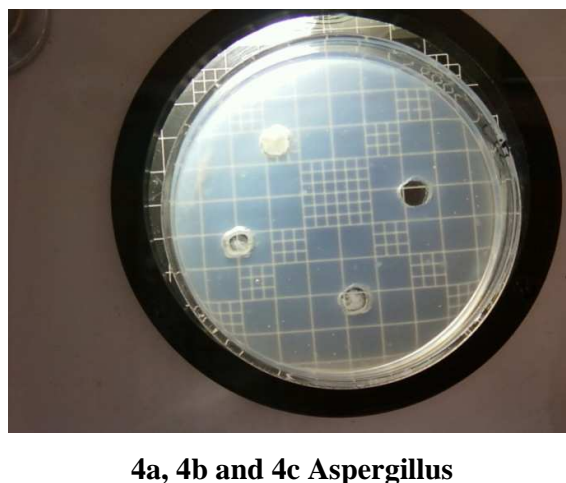
### Anti bacterial and Anti fungal activity:

All the synthesized compounds were screened of antimicrobial studies against antibacterial and antifungal activity by disc diffusion method and MIC by serial dilution method. All the synthesized compounds were subjected to preliminary antibacterial screening by disc diffusion method against *Bacillus subtilis* ATCC 6633 (gram positive) and *Escherichia coli* ATCC 11229 (gram negative). In this, Substituted-4-hydroxyquinolin-2(1H)-one **4a-c** series did not showed antibacterial activity against all tested organism *Escherichia coli* ATCC 11229 (gram negative) at the concentration of 50 µg/ml, 100 µg/ml and 150 µg/ml solutions (Table 1) (Figure. 2). In this Substituted-4-hydroxyquinolin-2(1H)-one **4a-c** series did not showed antifungal activity against all tested organism *Aspergillus Niger* MTCC F2723 at the concentration of 50 µg/ml, 100 µg/ml and 150 µg/ml solutions. (Table-2)(Figure-3).

**Figure 2: Antibacterial activity by disc diffusion method.**



**Figure 3: Antifungal activity by disc diffusion method.**



**Table-1: Antibacterial activity by disc diffusion method for 7- Substituted-4-hydroxyquinolin-2(1H)-one 4a-c:**

S.No	Comp	R	R1	Zone of inhibition (mm)					
				1. Bacillus subtilis			2. Escherichia coli		
				50µg	100µg	150µg	50µg	100µg	150µg
1	4a	F	H	3mm	5mm	6mm	NS	NS	NS
2	4b	Cl	H	3mm	4mm	5mm	NS	NS	NS
3	4c	Br	H	3mm	5mm	6mm	NS	NS	NS

**Zone of inhibition for standard samples:** Bacillus subtilis (11mm) and Escherichia coli (10mm).

**Abbreviation:** NS= Not significant.

**Table-2: Antifungal activity by disc diffusion method for 7- Substituted-4-hydroxyquinolin-2(1H)-one 4a-c:**

S.No	Compd	R	R1	Zone of inhibition (mm)		
				Aspergillus Niger		
				50µg	100µg	150µg
1	4a	F	H	NS	NS	NS
2	4b	Cl	H	NS	NS	NS
3	4c	Br	H	NS	NS	NS

**Zone of inhibition for standard samples:** Aspergillus Niger (08mm).

**Abbreviation:** NS= Not significant

**Conclusion:**

We have synthesized some novel 4-Hydroxy-1*H*-quinolin-2-ones derivatives. The newly synthesized 4-Hydroxy-1*H*-quinolin-2-ones derivatives are characterized by spectral data and further evaluated for antimicrobial activity. All compounds show moderate to good activity.

**References:**

1. Sudha B. S, Shashikant S, Khanum S.A & ShriharshaS N, Indian J. Pharm. Sci., **2003**: 65(5), 465-470.
2. Khan M.S.Y & Akhtar M., Indian journal of Chemistry, 2003, 42B, 900-904.
3. Peesapati Venkateswarlu & Srikant BVenkata Chitty, Indian journal of Chemistry, **2003**: 42B, 616-620.
4. R.Gill, R. Hargreaves and J.A. Kemp, J. Cereb. Blood Flow Metab 1991: 11, 304.
5. P.D. Leeson, B.J. Williams, M.Rowley, K.W. Moore, R. Baker and J.A. Kemp, Bioorg. Med. Chem. Lett., 1991:3, 71
6. C. F. Neviile, M.F. Grundon, V.N Ramachandran, g. Reish and J. Reisch, J. Chem. Soc. Perkin Trans 2002: 1, 2261.
7. E.Ziegler and Th. Kappe, Angew. Chem., 76, 921 (1964); Angew. Chem., Int. Ed. Engl., 2001. 3, 754.
8. E.Ziegler and Th. Kappe and R. Salvador, Monatsh. Chem., 1963: 94, 453.
9. A.M. Mcleod, S. Grim wood, C. Barton, L. Bristow, K. Say well, G.R. Marshall and R.G. Ball, J.Med. Chem., 1995:38, 2239
10. R.W. Carling, P.D. Leeson,K.W Moore, J.D. Smith, C.R. Moyes, I.M. Mower, S.Thoma, T.Chan, R.Baker, A.C. Foster, S. Grimwood, J.A Kemp, G.R . Marshal, M.T. Tricklebank and K.L. Say well , J. Med. Chem., 1993. 36, 3386.
11. P.Baumgarten and W. Kargel, Ber. 60, 832 (1927); improved method; W. Stadlbauer, O. Schmut5 and Th. Kappe, Monatsh. Chem., 1980.11, 1005.
12. D.R. Buckle, B.C.C. Cantello, H. Smith and B.A. Spicer, J. Med. Chem., 1975.18, 726.
13. Farbenfabrik Hoechst, D.R.P. 102, 894; Chem. Zblt., 1899 1, 462.
14. Badische Anilin und Sodafabrik, D. R. P. 117, 167; Chem. Zblt., 1901.1, 236.
15. C. M. Mehta and G. H. Patel, J. Ind. Res. (India), 1959 .18b, 391.

**Corresponding Author:**

**Subbaramireddy SR\***,

**Email:** [srsubbaramireddy@gmail.com](mailto:srsubbaramireddy@gmail.com)